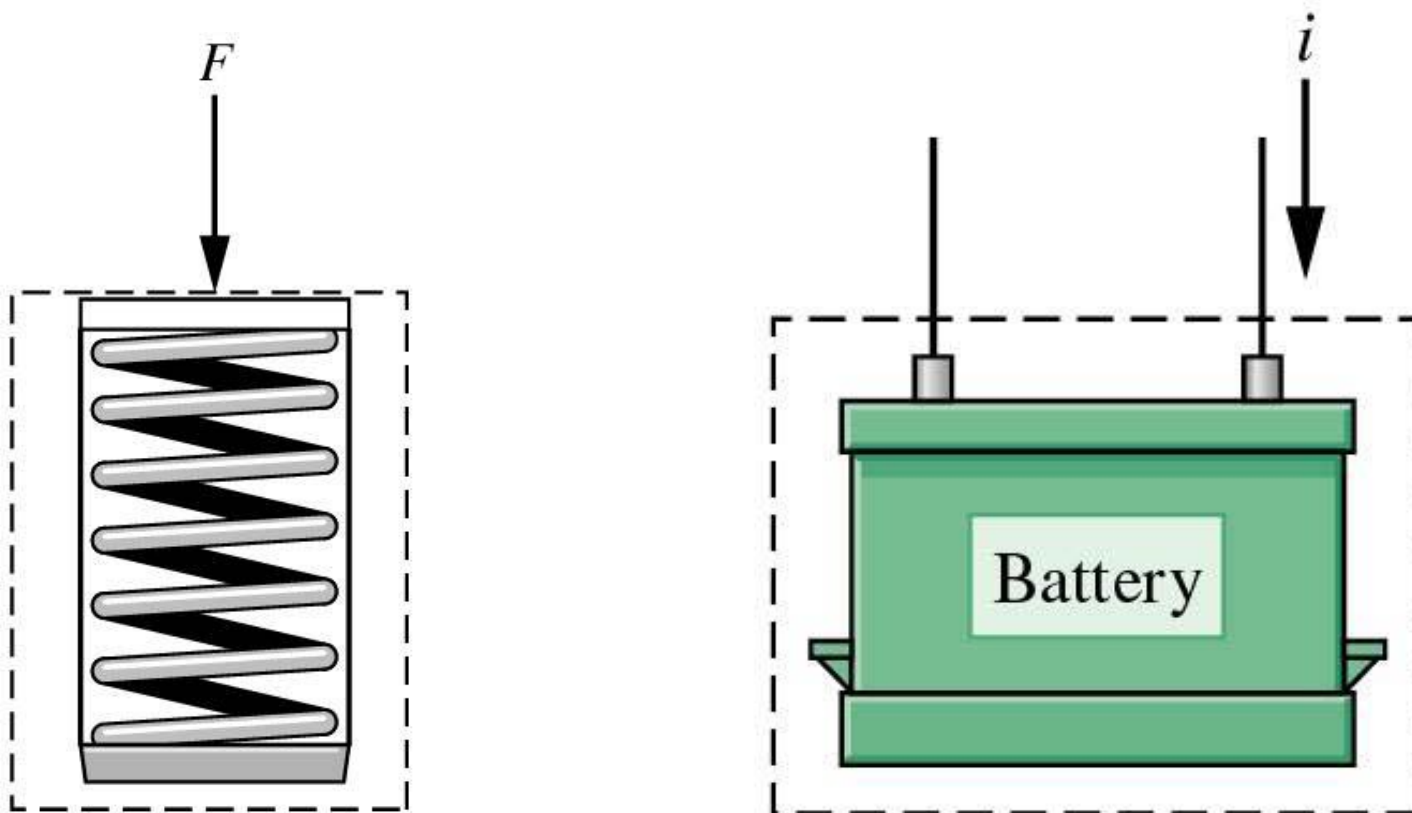


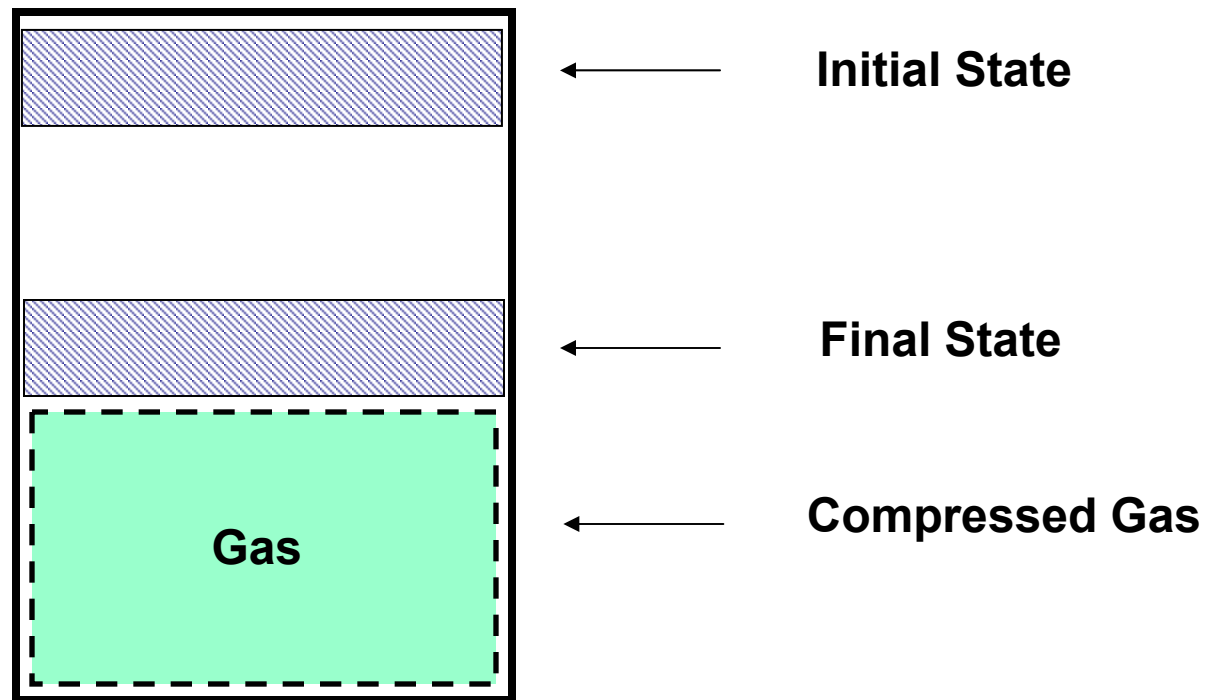
We will cover the following concepts,

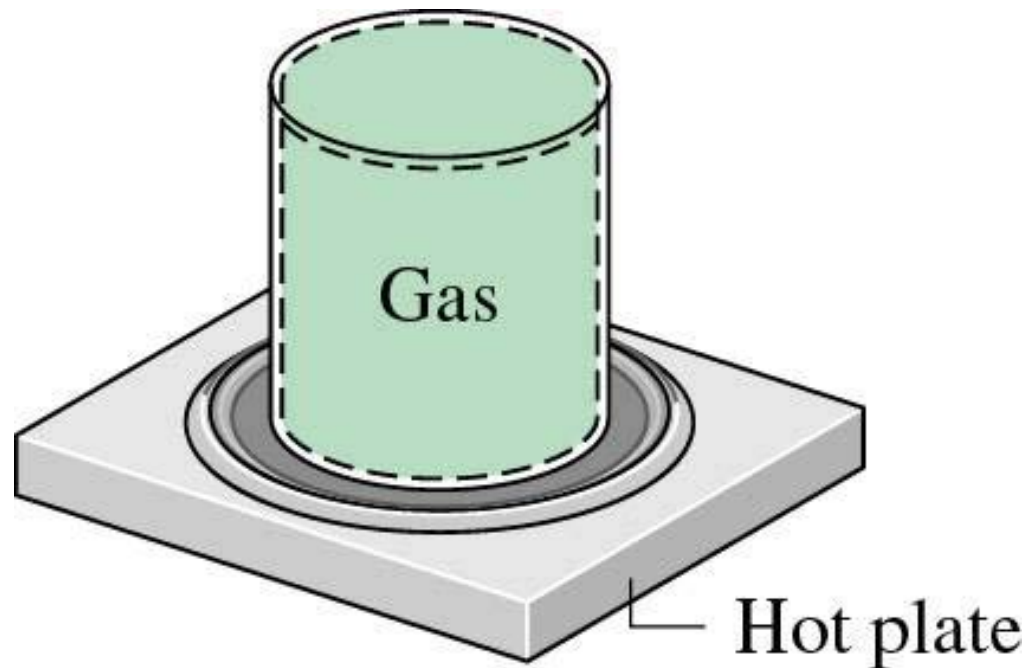
- ***Internal Energy***
- ***Energy Transfer by Heat***
- ***Heat Transfer Modes***
- ***First Law of Thermodynamics for a Closed System***

Internal Energy: Summation of any other forms of energy stored in a system is called Internal Energy



$$\begin{aligned}\Delta E &= \Delta KE + \Delta PE + \Delta U \\ &= (KE_2 - KE_1) + (PE_2 - PE_1) + (U_2 - U_1)\end{aligned}$$





$Q > 0$: heat transfer *to* the system

$Q < 0$: heat transfer *from* the system

Rate Of Energy Transfer By Heat

$$\dot{Q} = \frac{\text{Heat}}{\text{Time}} = \frac{dQ}{dt}$$

$$Q = \int_{t_1}^{t_2} \dot{Q} dt$$

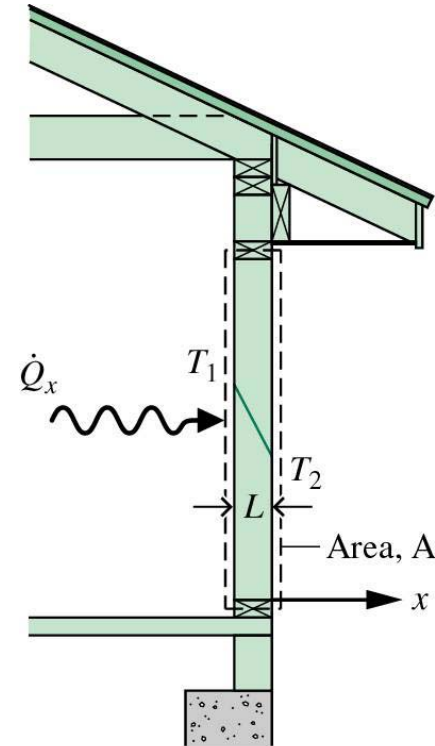
Heat Transfer Modes:

- **Conduction: Energy Transfer between the Particles of a Substance**
- **Radiation: Change in Electronic Configuration of the Molecules**
- **Convection: Energy Transfer when Contacting Solid-Fluid or Fluid-Fluid**

Fourier's Law:

$$\dot{Q}_x = -kA \frac{dT}{dx}$$

k is called conductivity.
 A is surface area.



- Linear Variation

$$\dot{Q}_x = -kA \frac{T_2 - T_1}{L}$$

Boltzmann's Law:

$$\dot{Q}_e = \varepsilon \sigma A T_b^4$$

ε emissivity is a property of the surface that shows how effectively the surface radiates.

σ is called Boltzmann's constant.

Newton's Law of Cooling

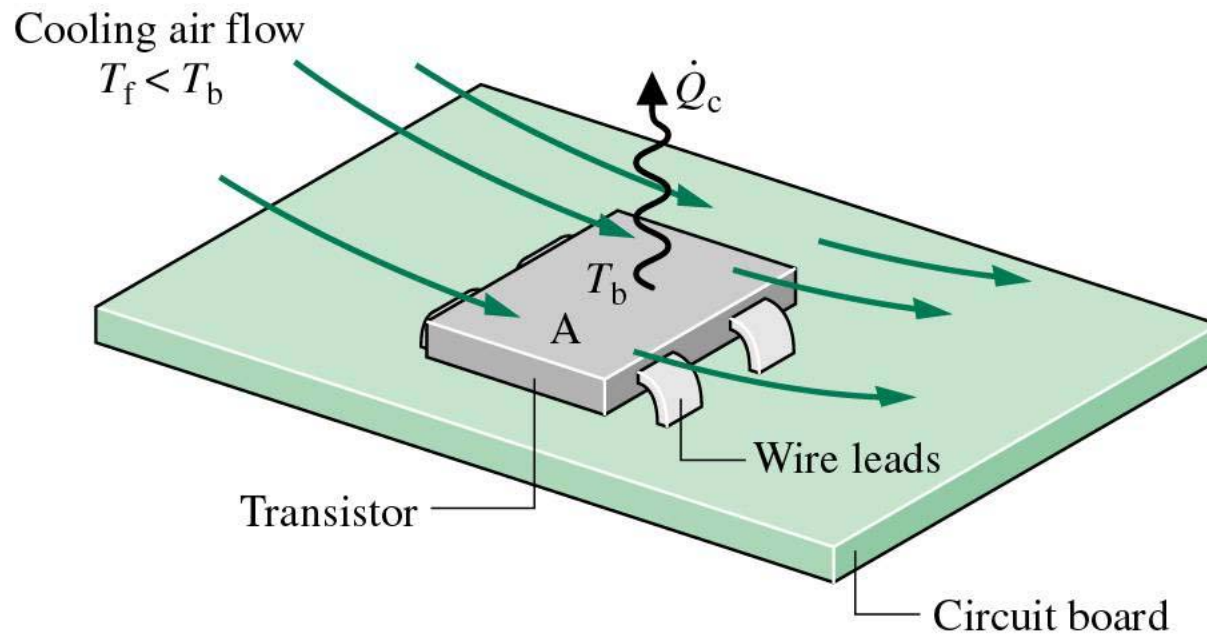
$$\dot{Q}_c = hA(T_b - T_f)$$

T_b is the solid surface temperature

T_f is the moving fluid temperature

h is the heat transfer coefficient

A is the surface area



Change in Total Energy = Net Heat Transfer In – Net Work Out

$$E_2 - E_1 = \Delta KE + \Delta PE + \Delta U = Q - W$$

Or

$$dE = \delta Q - \delta W$$

$W > 0$: work done *by* the system

$W < 0$: work done *on* the system

$Q > 0$: heat transfer *to* the system

$Q < 0$: heat transfer *from* the system

**Time Rate of Change of Total Energy =
Net Rate of Heat Transfer *In* – Net Rate of Work *Out***

$$E_2 - E_1 = \Delta KE + \Delta PE + \Delta U = Q - W$$

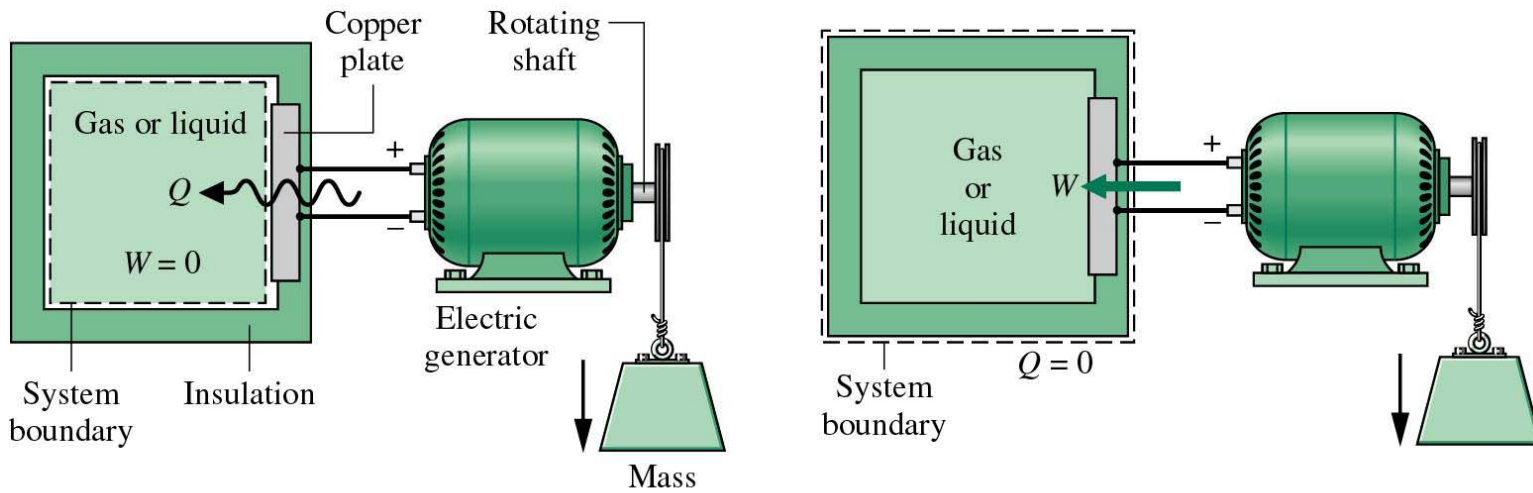
Or

$$\frac{dE}{dt} = \frac{dKE}{dt} + \frac{dPE}{dt} + \frac{dU}{dt} = \dot{Q} - \dot{W}$$

Specific Properties

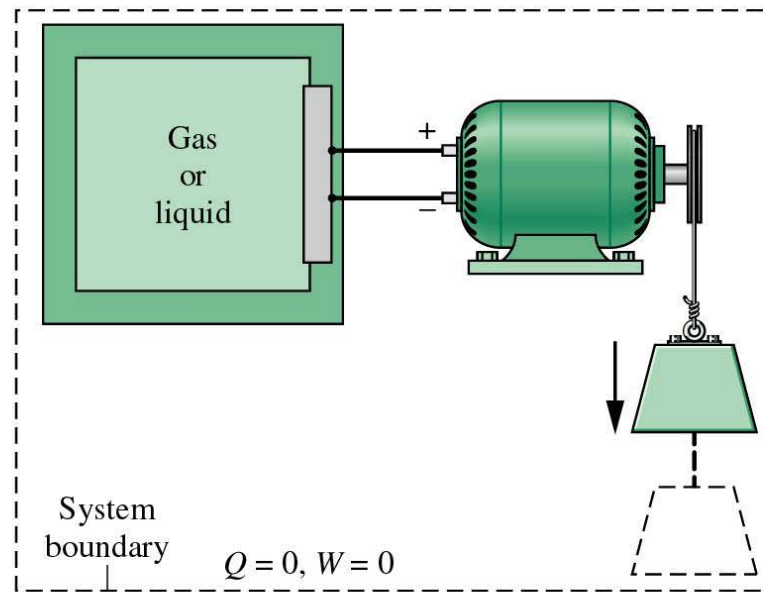
$$e = \frac{E}{m}, u = \frac{U}{m}$$

First Law of Thermodynamics



(a)

(b)



(c)