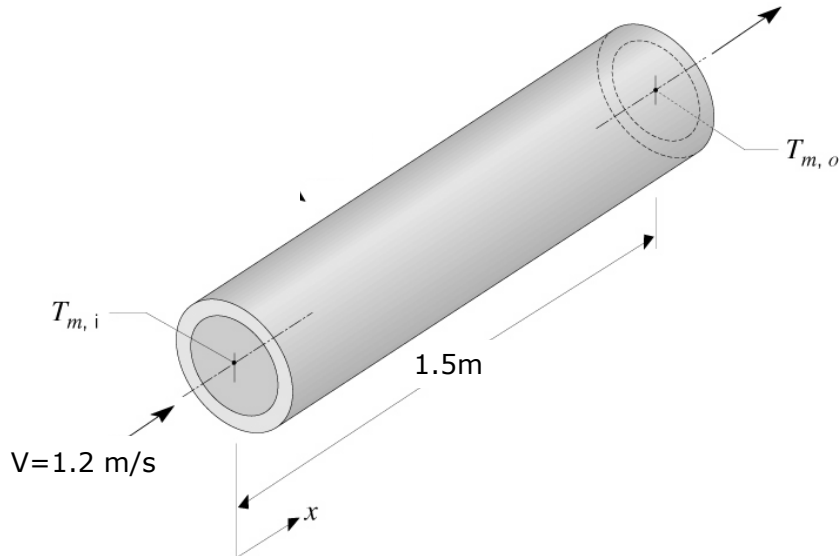


Name:

(1 Hour)

If solutions involve an iterative process multiple iterations are not necessary.

Problem 1 (10 points): Nitrogen gas at atmospheric pressure and bulk inlet temperature of 27°C is heated in a tube with inner diameter of 2.5 cm at a constant surface temperature of 100°C. The bulk average velocity is 1.2 m/s and the tube is 1.5m long. Determine the net heat transfer rate to the nitrogen.



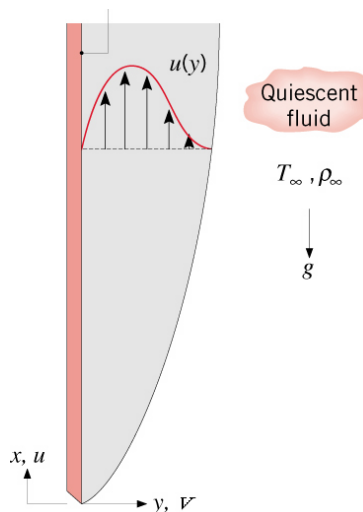
Problem 2 (10 points): One side of a plate of dimensions 6x6x0.01m receives radiant heat flux from the sun of 1100 w/m². Assuming that 95 w/m² is conducted from the opposite side of the plate, estimate the *mid* surface temperature of the plate if the atmospheric air is at T_∞=20°C, and the plate is

(a) vertical,

(10 points)

(b) **(Bonus)** inclined $\theta=45^\circ$.

(3 points)



Problem 1

$$\overline{T}_{m,i} = \overline{T}_i = 15^\circ\text{C} \quad d_i = 2.5 \text{ cm}$$

$$V = 1.2 \text{ m/s} \quad T_s = 100^\circ\text{C}$$

$$L = 1.5 \text{ m}$$

The exit temperature is unknown so we need to guess some value for it, and then correct our guess.

$$\text{Trial 1: } T_{m,o} = 53^\circ\text{C} \rightarrow \overline{T}_m = \frac{27 + 53}{2} = 40^\circ\text{C} = 313 \text{ K}$$

Table A4 (Nitrogen): $\rho = 1.105 \text{ kg/m}^3$, $\nu = 16.94 \times 10^{-6} \text{ m}^2/\text{s}$
 $C_p = 1.042 \times 10^3 \text{ J/kgK}$, $Pr = 0.710$
 $k = 0.0271 \text{ W/mK}$
 $\mu = 18.37 \times 10^{-6} \text{ kg/ms}$, $M_s = 20.86 \times 10^{-6} \text{ kg/m}^3$
 \rightarrow at $T_s = 100^\circ\text{C}$

check the flow regime

$$Re_D = \frac{DV}{\nu} = \frac{0.025 \times 1.2}{16.94 \times 10^{-6}} = 1771 < 2300 \text{ Laminar}$$

now lets check if it is fully developed or not

$$\text{Velocity: } X = 0.05 D Re_D = 0.05 \times 0.025 \times 1771 = 2.215 \text{ m} > 1.5 \text{ m}$$

$$\text{Temperature: } X_+ = 0.05 D Re_D Pr = 0.05 \times 0.025 \times 1771 \times 0.71 = 1.572 \text{ m} > 1.5 \text{ m}$$

so both velocity & temperature profiles are still developing,

using the Eq. 8.57 for entry region we have

$$\overline{Nu}_D = 1.86 \left(\frac{Re_D Pr}{L/D} \right)^{1/3} \left(\frac{\mu}{\mu_s} \right)^{0.14}$$

$$\text{So } \overline{Nu_D} = 1.86 \left(\frac{1771 \times 0.710}{60} \right)^{1/3} \left(\frac{18.37}{20.86} \right)^{0.14} = 5.037$$

Then

$$\bar{h} = \overline{Nu_D} \frac{k}{D} = 5.037 \times \frac{0.0271}{0.025} = 5.46 \text{ W/m}^2\text{K}$$

Now we can calculate the $T_{m,o}$ using Eq. 8.42

$$\frac{T_s - T_{m,o}}{T_s - T_{m,i}} = \exp\left(\frac{-PL}{\dot{m}c_p} \bar{h}\right) \quad \text{for } T_s = \text{const.}$$

$$\text{where } \dot{m} = \rho VA = \rho V \left(\pi \frac{D^2}{4} \right) = 1.105 \times 1.2 \times \left(3.1415 \times \frac{0.025^2}{4} \right)$$

$$\rightarrow \dot{m} = 6.51 \times 10^{-4} \text{ kg/s}$$

Now we have

$$\frac{100 - T_{m,o}}{100 - 27} = \exp\left(\frac{-\overbrace{\pi(0.025) \times 1.5 \times 5.46}^P}{6.51 \times 10^{-4} \times 1041.5}\right)$$

This gives

$$T_{m,o} = 71.7^\circ\text{C}$$

This is the corrected outlet temperature

to obtain a better result we need to iterate once more,

$$\text{Trial 2: } T_{m,o} = 71.7^\circ\text{C} \rightarrow \bar{T}_m = \frac{27 + 71.7}{2} = 49.4^\circ\text{C} \approx 50^\circ\text{C} = 323\text{K}$$

Table A4 (Nitrogen):

$$\rho = 1.076 \text{ kg/m}^3$$

$$c_p = 1.0419 \times 10^3 \text{ J/kgK}$$

$$\nu = 17.95 \times 10^{-6} \text{ m}^2/\text{s}$$

$$k = 0.0278 \text{ W/mK}$$

$$Pr = 0.708$$

$$\mu = 18.74 \times 10^{-6} \text{ Pa}\cdot\text{s}$$

Again check the flow regime

$$Re_D = \frac{DV}{\nu} = \frac{0.025 \times 1.2}{17.95 \times 10^{-6}} = 1671 < 2300 \text{ laminar}$$

$$\text{now } \overline{Nu}_D = 1.86 \left(\frac{1671 \times 0.708}{60} \right)^{1/3} \left(\frac{18.79 \times 10^{-6}}{20.86 \times 10^{-6}} \right)^{0.14}$$

$$\rightarrow \overline{Nu}_D = 4.95 \rightarrow \bar{h} = \frac{k}{D} \overline{Nu}_D = \frac{0.0278}{0.025} (4.95)$$

$$\text{and } \rightarrow \bar{h} = 5.504 \text{ W/m}^2\text{K}$$

$$\dot{m} = \rho AV = 1.076 \times \pi \times \frac{0.025^2}{4} \times 1.2 = 6.33 \times 10^{-4} \text{ kg/s}$$

Then

$$\frac{100 - T_{m,o}}{100 - 27} = \exp\left(\frac{-\pi \times 0.025 \times 1.5 \times 5.504}{6.33 \times 10^{-4} \times 10419}\right)$$

Then

$$\boxed{T_{m,o} = 73 \text{ } ^\circ\text{C}} \text{ Ans.}$$

So this is sufficient for the $T_{m,o}$ evaluation

now to find the heat transfer we need

the log-mean temperature,

$$\Delta T_{lm} = \frac{\Delta T_o - \Delta T_i}{\ln(\Delta T_o / \Delta T_i)} = \frac{(100 - 73) - (100 - 27)}{\ln[(100 - 73) / (100 - 27)]} = 46.5^\circ\text{C}$$

$$\rightarrow q_{\text{conv.}} = \bar{h} A_s \Delta T_{lm} = 5.504 \times \pi \times 0.025 \times 1.5 \times 46.5^\circ\text{C}$$

$A_s = \pi DL$

$$\rightarrow q_{\text{conv.}} = 30.2 \text{ W}$$

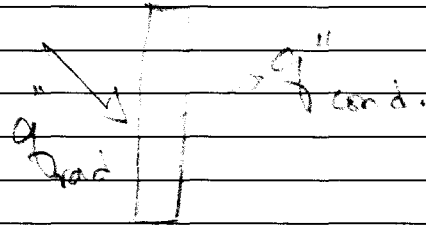
Problem 2.

$$L = W = 6 \text{ m}$$

$$\delta = 0.01 \text{ m}$$

$$q''_{\text{rad}} = 1100 \text{ W/m}^2$$

$$q''_{\text{cond.}} = 95 \text{ W/m}^2$$



$$T_{s, L/2} = T_s(x = \frac{L}{2}) = ?$$

Start with Energy Balance.

$$E_{\text{in}} - E_{\text{out}} = E_{\text{st}} - E_{\text{gen}}$$

$$\rightarrow q''_{\text{rad}} - q''_{\text{cond.}} = q''_{\text{system}} = q''$$

$$\rightarrow q'' = 1100 - 95 = 1005 \text{ W/m}^2$$

We can use equation 9.26 but need to use $T_{s, L/2}$ for the properties and we don't have this temperature, so we need to guess.

$$\text{Trial 1: } T_{s, L/2} = 75^\circ\text{C} \rightarrow T_p = \frac{75 + 20}{2} = 320.5 \text{ K}$$

$$\text{Table A4 (air): } \alpha = 25 \times 10^{-6} \text{ m}^2/\text{s}, \quad k = 28 \times 10^{-3} \text{ W/mK}$$

$$\nu = 17.95 \times 10^{-6} \text{ Pa.s}$$

$$Pr = 0.7$$

$$B = \frac{1}{T_p} = 3.12 \times 10^{-3} \text{ K}^{-1}$$

Now

$$Ra_L = \frac{g\beta(T_{s,L/2} - T_m)L^3}{\nu\alpha} = \frac{9.81 \times 3.12 \times 10^{-3} (75 - 20) \times 6^3}{17.95 \times 10^{-6} \times 25 \times 10^{-6}}$$

$$\rightarrow Ra_L = 8.1 \times 10^{11} > 10^9 \quad \text{Turbulent}$$

use Eq. 9.26

$$\overline{Nu}_L = \left[8.25 + \frac{0.387 Ra_L^{1/4}}{\left(1 + \left(\frac{0.492}{Pr}\right)^{4/16}\right)^{3/27}} \right]^2$$

$$\rightarrow \overline{Nu}_L = 1563.3 \rightarrow \bar{h} = \overline{Nu}_L \frac{k}{L} = 1563.3 \times \frac{28 \times 10^{-3}}{6}$$

$$\rightarrow \bar{h} = 7.296 \text{ W/m}^2\text{K}$$

Now we can find the correct $T_{s,L/2}$,

$$q'' = \bar{h} \Delta T_{L/2} = \bar{h} (T_{s,L/2} - T_m)$$

$$\rightarrow 1005 = 7.296 (T_{s,L/2} - 20) \rightarrow T_{s,L/2} = 157.75^\circ\text{C}$$

Now let's correct the results using this Temp.

$$\bar{T}_m = \frac{T_{s,L/2} + T_m}{2} = \frac{157.75 + 20}{2} = 88.9^\circ\text{C} = 361.9 \text{ K}$$

Table A9/air \rightarrow $k = 0.0313 \text{ W/mK}$, $\beta = 2.725 \times 10^{-3} \text{ K}^{-1}$
 $\nu = 22.8 \times 10^{-6} \text{ m}^2/\text{s}$
 $\alpha = 32.8 \times 10^{-6} \text{ m}^2/\text{s}$
 $Pr = 0.697$

check Ray again

$$Ra_L = \frac{9.81 \times 2.725 \times 10^{-3} \times (157.75 - 20) \times 6^3}{22.8 \times 10^{-6} \times 32.8 \times 10^{-6}} = 1.064 \times 10^{12} > 10^9$$

Turbulent

now Eq. (9.26)

$$Nu_L = \left[0.25 + \frac{0.387 (1.064 \times 10^{12})^{1/6}}{\left[1 + \left(\frac{0.492}{0.697} \right)^{9/16} \right]^{3/27}} \right]^2$$

$$\rightarrow Nu_L = 1679.50 \rightarrow \bar{h} = Nu_L \frac{k}{L} = 1679.50 \times \frac{0.0313}{6}$$

$$\rightarrow \bar{h} = 8.76 \text{ W/m}^2\text{K}$$

and

$$q'' = \bar{h} \Delta T_{L/2} = \bar{h} (T_{s,L/2} - T_\infty)$$

$$\rightarrow 1005 = 8.76 (T_{s,L/2} - 20)$$

$$\rightarrow \boxed{T_{s,L/2} = 134.71 \text{ }^\circ\text{C}} \quad \text{for Vertical Wall}$$

Bonus:

in $Ra_L = \frac{g \beta \Delta T L^3}{\nu \alpha}$ replace "g" with "g cos θ "

so $Ra_L = \frac{\beta g \cos \theta \Delta T L^3}{\nu \alpha}$ and repeat the

problem.